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compounds. 12.. 16. 17. 19. 20. So Cac IGde Chemistry

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enstructional Fair, Inc. 5. 35 g of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ Determine the
number of grams in each of the quantities below. 1, 2.5 moles of
NaCl 2. 0.50 moles of 3, 1.70 moles of KMnO_4 4, 0.25 moles of

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|He SAI

A chemical equation is the symbol in Chemistry that represents chemical reaction with the help of chemical formulas. It contains the chemical substances that are involved in the reaction. It contains reactants and products. The reactants are the elements that react with one and another in a chemical reaction, while the products are the elements ...

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Shift none Same 10,6 kcal Amount NaOH(s) Same Same 82

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decreases same Same remains the increases same same same
Same Same same same same (Remember that pure solids and
liquids do not affect equilibrium values.) $[Na^+]$ same Sa $[OH^-]$
same Same same Sa me same same sanqe

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